

工學碩士學位論文

2002年 2月

釜慶大學校大學院

食品工學科

趙 常 元

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指導教授 金 善 奉

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Deodorizing Activity of Apple Extracts

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Abstract

Polyphenol oxidase extracted from apple, *Malus pumila*(Hongro), were investigated by measuring the changes of methyl mercaptan as an indicator of halitosis in human mouths in order to know the deodorizing activity on the added the amounts of apple extracts, their molecular fractions, substrate specificity, pH, reducing agents, oxygen and metal ions.

The deodorizing activity of apple extracts was increased with the amount of apple extracts. The deodorizing activity on adding polyphenol oxidase to the low molecular fraction of apple extracts was increased with the amount of the low molecular fraction of apple extracts and the reaction time of the extracts with polyphenol oxidase.

The deodorizing activity on substrate specificity of polyphenol

oxidase showed the high affinity of 81.3% with chlorogenic acid and 36.8% with catechol.

The deodorizing activity of polyphenol oxidase on pH changes against methyl mercaptan was 20% at below at pH 5.5 and 80% at above pH 5.5.

When the reducing agent was added to polyphenol oxidase the deodorizing effect of polyphenol oxidase against methyl mercaptan decreased. It was assumed that the reducing agent prevented the oxidation of chlorogenic acid.

The deodorizing activity of polyphenol oxidase against methyl mercaptan required oxygen. The deodorizing effect of polyphenol oxidase from apple against methyl mercaptan was 20% in an anaerobic state.

When Cu^{2+} or Fe^{3+} was added to polyphenol oxidase the deodorizing effect of polyphenol oxidase against methyl mercaptan increased. It was assumed that metal ions promoted the production of *o*-quinone.

The deodorizing activity of polyphenol oxidase is thought that *o*-quinone as an intermediate produced by an oxidative reaction of polyphenol oxidase during enzymatic browning reactions may react with methyl mercaptan to form a kind of non-volatile and sulfur-containing compound.

(口臭)

가

가

가

(Spouge, 1964).

가

가 . 가

, 가 가
(Uchida, 1974).

, ,
, aldehyde ,
hydrogen sulfide, methyl mercaptan, dimethyl sulfide
methyl mercaptan
가
methyl mercaptan
(Tonzetich et al., 1964; Tsunoda et al., 1975).

,
epigallocatechin gallate(EGCg) 가
(Sato and Ishikawa, 1984;
Suzuki and Uchida, 1984). EGCg methyl mercaptan
EGCg가 o-quinone methyl
mercaptan thiol group methyl mercaptan
(Yasuda and Arakawa, 1995).

flavonoid, 가
(Kaji, 1976; Tokita et al., 1984; Urabe et al.,
1999; Miura, 1989; Yasuda and Ui, 1992). Lee et al.(1999,
2000, 2001) 28 (7,

12, 9) methyl mercaptan
90% methyl mercaptan

Streptococcus mutans glucosyltransferase

polyphenol

oxidase(E.C. 1. 10. 3. 1) polyphenol
o-quinone

polyphenol oxidase

, pH, , EDTA

1.

1. 1.

, *Malus pumila*(),

4 .

1. 2.

pH 3.0 5.0 0.01M citrate-0.02M
phosphate, pH 6.0 8.0 0.01M Na₂HPO₄-NaH₂PO₄

1. 3.

methyl mercaptan Wako,
L-tyrosine, catechol, chlorogenic acid, L-3,4-dihydroxyphenylalanine
(L-DOPA) gallic acid sigma .

1. 4.

300 g ethanol 가 80%
4°C 2 . 4°C 3000 rpm

30

-80 °C

1. 5. Methyl mercaptan

Methyl mercaptan (1 $\mu\text{g}/\mu\text{l}$ in benzene ; Wako Pure Chem., Osaka, Japan) 2 Ml 198 Ml ethanol

4

10 (1 $\mu\text{g}/\text{Ml}$)

2.

2. 1. Polyphenol oxidase

Polyphenol oxidase Flurkey et al.(1978)

200 g -23°C acetone 300 Ml 가 2

Whatman #1

-23°C acetone 150 Ml 가

Whatman #1

3 acetone powder

acetone powder 2 g 1 M KCl 0.05

M phosphate buffer(pH 6.5) 200 Ml 가 4°C 30

4°C 10,000 rpm 15

glass wool

polyphenol oxidase

2. 2. Polyphenol oxidase

Polyphenol oxidase Zenin et al.(1978)
10mM catechol 0.01M
sodium phosphate buffer(pH6.5) 2.9ml cell water bath
30°C , 0.1ml 가 ,
420nm (Shimazu, UV-160A) , 1
1ml 0.01 가 1unit .

2. 3.

polyphenol oxidase
가 37 6 incubate 420nm
.

2. 4.

Tokita et al.(1984) Fig. 1
0.2M potassium phosphate
buffer(PPB) 1M 30M vial pH 7.5
methyl mercaptan (1µg/M) 1M 가
vortex mixer 5
37 6 incubate vial headspace methyl
mercaptan FPD(flame photometric detector)가 gas
chromatography(GC) ,
.

$$(\%) = \frac{C - S}{C} \times 100$$

C : control methyl mercaptan

S : 가 methyl mercaptan

2. 5. Gas chromatography(GC)

Methyl mercaptan FPD(flame photometric detector)가 GC(Hewlett Packard 5890, series ; Column, HP-1 phase(5m × 0.53mm × 2.65μm) ; Column temperature, 35 ; Injector temperature, 150 ; Detector temperature, 200 ; Carrier Gas(He) flow rate, 40Mℓ/min) (Table 1).

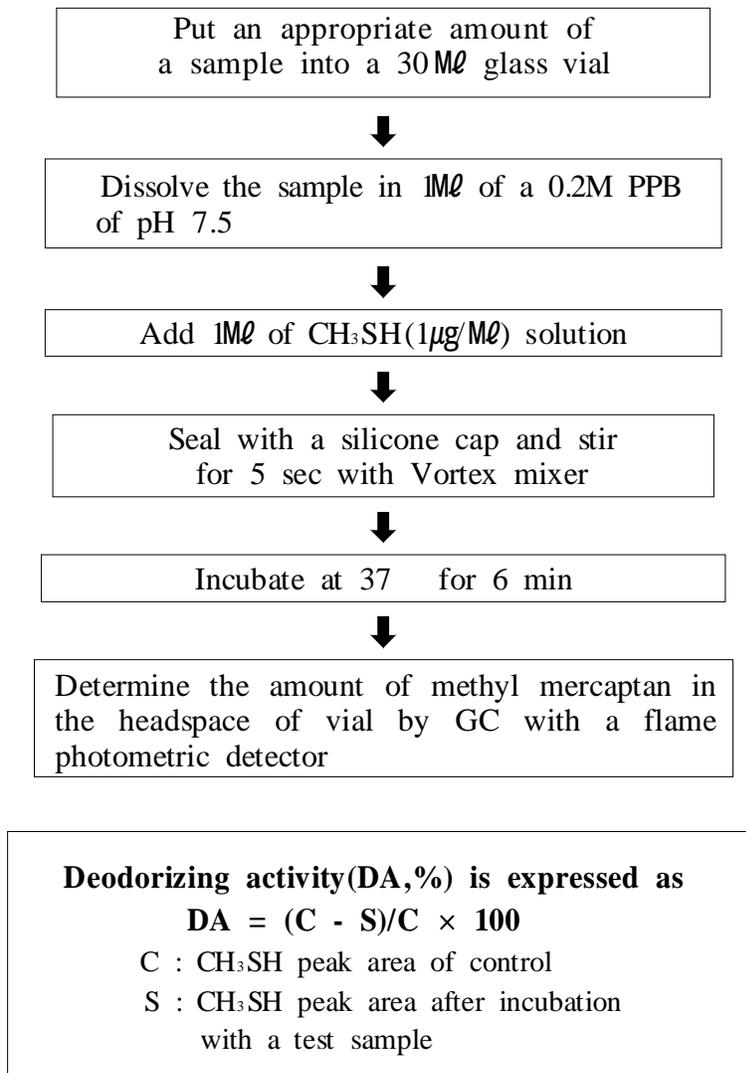


Fig. 1. Flow chart for the deodorizing assay.

Table 1. Operating condition of gas chromatography for measuring deodorizing activities

Gas chromatograph	He wlett Packard 5890, series
Column	HP-1(methyl silicone gum)phase, 5m×0.53mm×2.65 μ m
Column temperature	35°C
Injector temperature	150°C
Detector temperature	200°C
Detector	Flame photometric detector(FPD)
Carrier gas	He : 40 <i>Me</i> /min

1.

methyl mercaptan

Fig. 1

가 가 가
, 10 mg/Mℓ 가 73.5%
. Lee et al.(1999)

sodium fluoride, cetylpyridinium

chloride, benxethonium methyl mercaptan
5 mg/Mℓ 50%

2. Polyphenol oxidase

pH

polyphenol oxidase

pH

Fig. 3 4

pH 3.0 0.74% pH 가 가
가 pH 6.5 가
pH

Chung et al.(1983, 1984)

polyphenol oxidase

pH가 6.5, 6.0 , Choi et al.(1987)

pH가 5.5 . Rivas and Whitaker(1973), Luh and

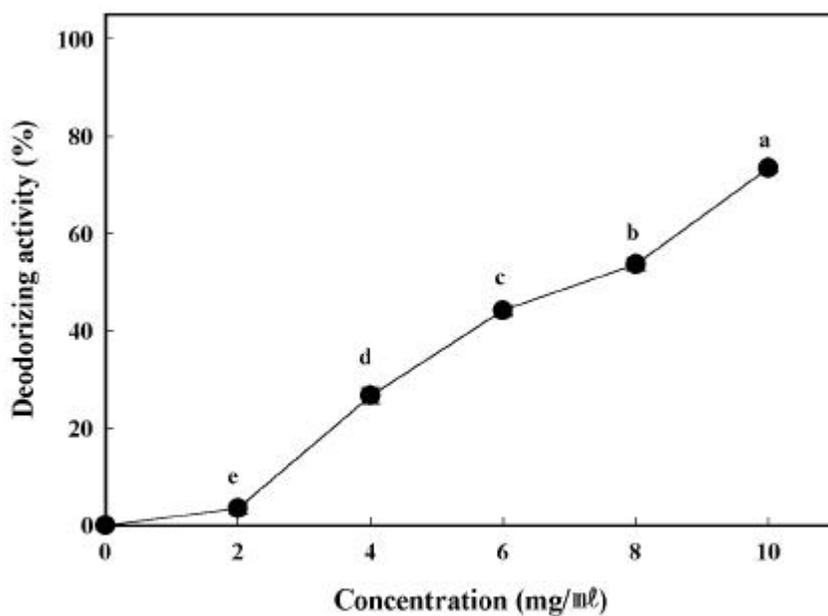


Fig. 2. Effect of the concentration of apple extracts on the deodorizing activity.

Values are means of three replicates and those with different letters are significantly different at $p < 0.05$.

Phithapol(1972) Park et al.(1980) , , Mango
 pH가 6.2, 6.2, 5.6 6.0 .
 가
 가 10°C 25%
 가 가 30°C
 가 30°C
 60°C 20% .
 Wissemann and Lee(1981) polyphenol oxidase
 25°C , Ha and Lee(1988) polyphenol
 oxidase 37°C .
 , polyphenol oxidase .

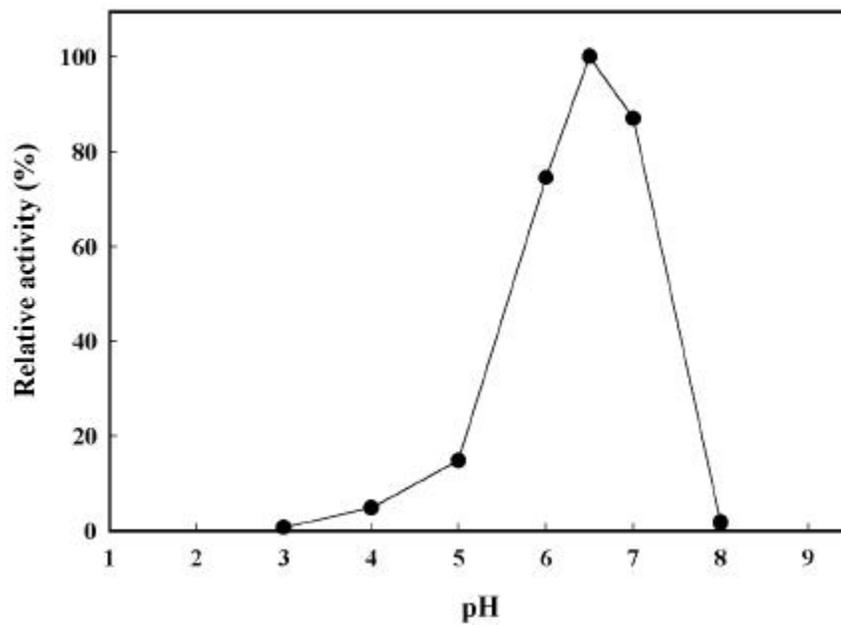


Fig. 3. Effect of pH on the activity of polyphenol oxidase from apples.

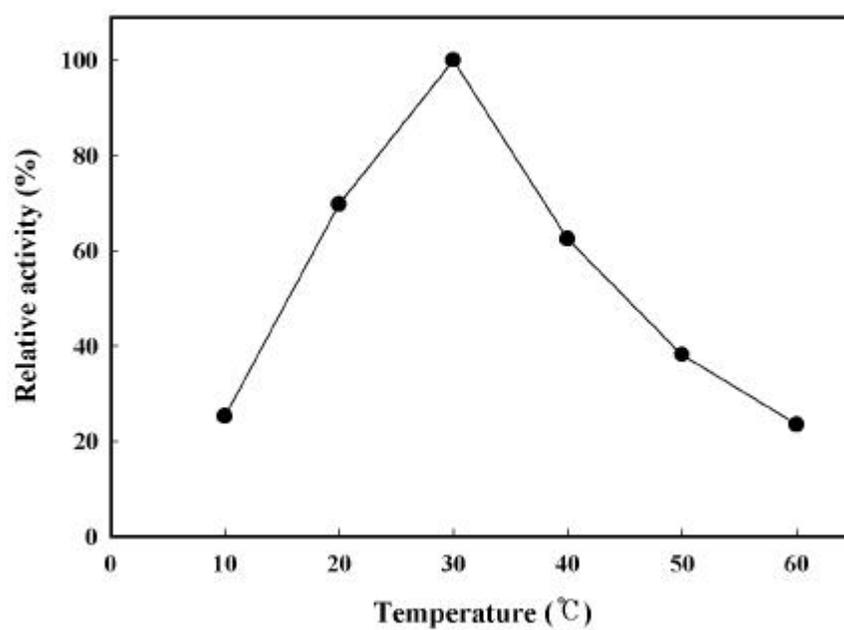


Fig. 4. Effect of temperature on the activity of polyphenol oxidase from apples.

3. **oxidase** **polyphenol**

() () ,

polyphenol oxidase 가

Fig. 5 Fig. 6 .

Polyphenol oxidase 가

가 가

, 가

가 (P<0.05).

polyphenol oxidase phenol

phenol

phenol

chlorogenic acid, catechin, procyanidin B₂, procyanidin C₁, phloridzin, l-epicatechin, phloretin xylogucoside

(Wieslaw et al., 1988).

polyphenol oxidase 가 (P<0.05).

Phenol polyphenol oxidase

가 methyl mercaptan thiol 가

. (2001) cysteine, glutathione

SH

polyphenol oxidase

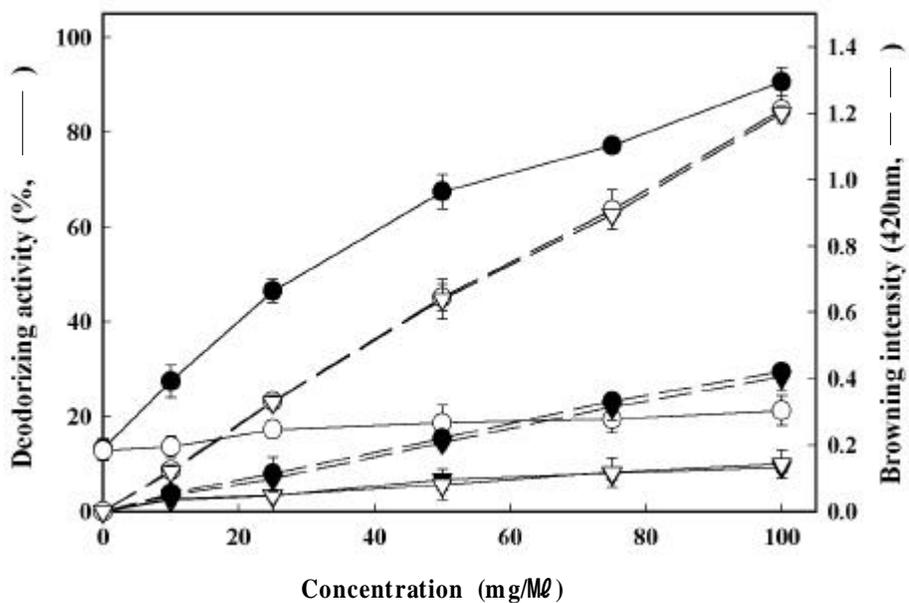


Fig. 5. Deodorizing activity and browning intensity of the added polyphenol oxidase on the concentration of supernatants and residues by 80% ethanol treatment of apple extracts.

- ; polyphenol oxidase-added supernatants
- ; polyphenol oxidase-added residues
- ▼ ; supernatants without adding polyphenol oxidase
- ▽ ; residues without adding polyphenol oxidase

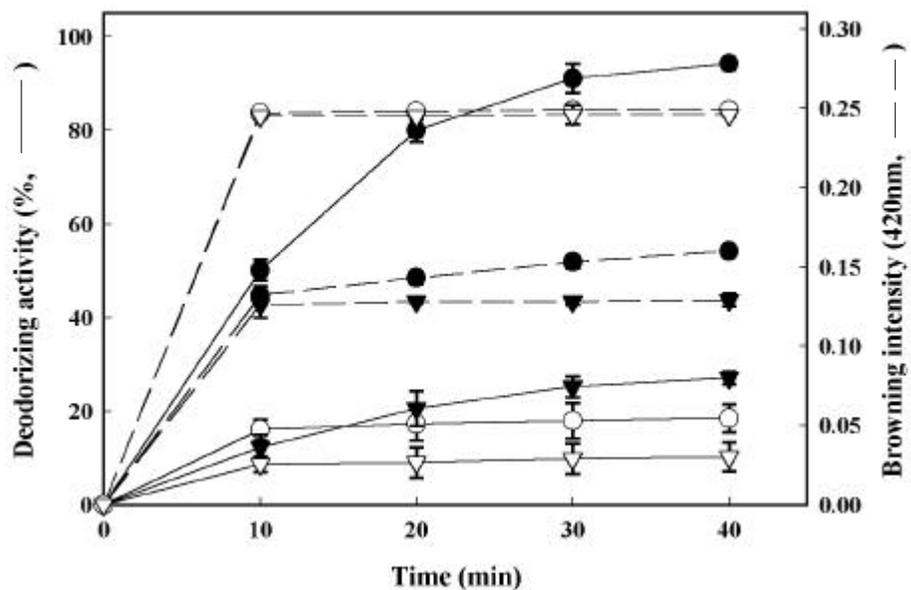


Fig. 6. Effect of reaction time of polyphenol oxidase on deodorizing activity and browning intensity in supernatants and residues by 80% ethanol treatment of apple extracts.

- ; polyphenol oxidase-added supernatants
- ; polyphenol oxidase-added residues
- ▼ ; supernatants without adding polyphenol oxidase
- ▽ ; residues without adding polyphenol oxidase

phenol

.

,

polyphenol oxidase 가

가

가 (P<0.05).

polyphenol oxidase 가

가

(p<0.05).

polyphenol oxidase

o-quinone methyl mercaptan

methyl mercaptan

. Yasuda et al.(1995)

EGCg methyl mercaptan

EGCg가

o-quinone methyl mercaptan thiol group

methyl mercaptan

.

4. Polyphenol oxidase

	polyphenol oxidase		methyl mercaptan
monophenol	L-tyrosine, <i>o</i> -diphenol	catechol, chlorogenic acid,	
L-DOPA	trihydroxyphenol	gallic acid	

Table 2

	Polyphenol oxidase		chlorogenic acid	catechol
			<i>o</i> -diphenol	
	,		polyphenol oxidase	
(Jung et al., 1983).		<i>o</i> -diphenol	chlorogenic acid	
가			polyphenol oxidase	
		chlorogenic acid	Walker et al.(1965)	
		, Jung et al.	(1984)	(1983)
				polyphenol oxidase
	esculetin	dihydroquercetin		<i>l</i> -epicatechin
	(Shmsnnon and Pratt, 1967).			
			polyphenol oxidase	
				chlorogenic acid
catechol	81.3%	36.8%		.

Table 2. Effect of polyphenol compounds on the deodorizing activity of polyphenol oxidase(PPO) from apples

	Substrate (10mM)	Activity (unit)	Relative activity (%)	Deodorizing activity (%)
Without PPO	L-tyrosine	0	-	0
	Chlorogenic acid	0	-	2.5
	Catechol	0	-	6.8
	L-DOPA ^a	0	-	0
	Gallic acid	0	-	0
With PPO	L-tyrosine	0	0	0
	Chlorogenic acid	78.0	205.3	81.3
	Catechol	38.0	100.0	36.8
	L-DOPA	6.7	17.6	13.9
	Gallic acid	1.3	3.4	13.6

^aL-DOPA means L-3,4-dihydroxyphenylalanine.

5.

5. 1. pH

10mM chlorogenic acid pH가
 polyphenol oxidase pH가

Fig. 7

polyphenol oxidase methyl mercaptan
 pH가
 pH 7.5 86.5%, pH 5.5
 14.7% 가

. Yasuda and Arakawa(1995) catechin
 가 pH 5.0

. Lee et al.(2000)
 pH가

, Mcnamara et al.(1972) 가
 가 37°C 24 ,
 가
 , 가 pH가
 가 가

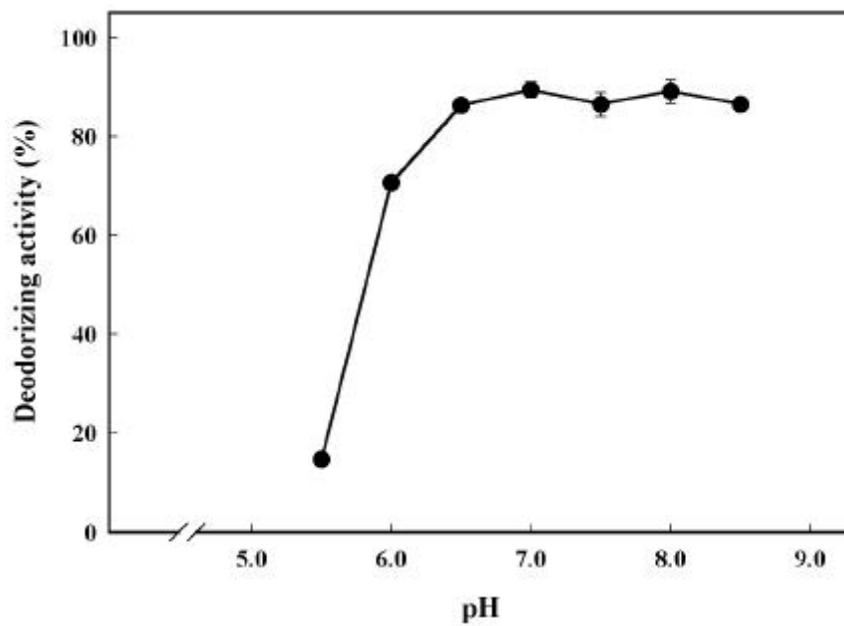


Fig. 7. Effect of pH on the deodorizing activity of polyphenol oxidase from apples against methyl mercaptan.

5. 2.

polyphenol oxidase

ascorbic acid sodium hydrosulfite

가 Fig. 8 .

Ascorbic acid sodium hydrosulfite 가 가

. Yasuda and Onogi(1996)

EGCg ascorbic acid 가

가

, Lee et al.(2000)

. Mcevely et al.(1992) o-diphenols polyphenol oxidase,

O₂ o-quinones

ascorbic acid sulfites

.

Fig. 9 ascorbic acid 가가 polyphenol oxidase methyl mercaptan

.

polyphenol oxidase 20μℓ, 10mM chlorogenic acid 20μℓ, 0.2M potassium phosphate buffer(pH 7.5)

1Mℓ methyl mercaptan (0.01μg/Mℓ) 1Mℓ 30Mℓ

vial ascorbic acid 가 ascorbic acid(50μg/Mℓ) 1Mℓ

가 가 1Mℓ 가

37°C 6, 10, 15, 20

headspace gas 100μℓ sampling GC .

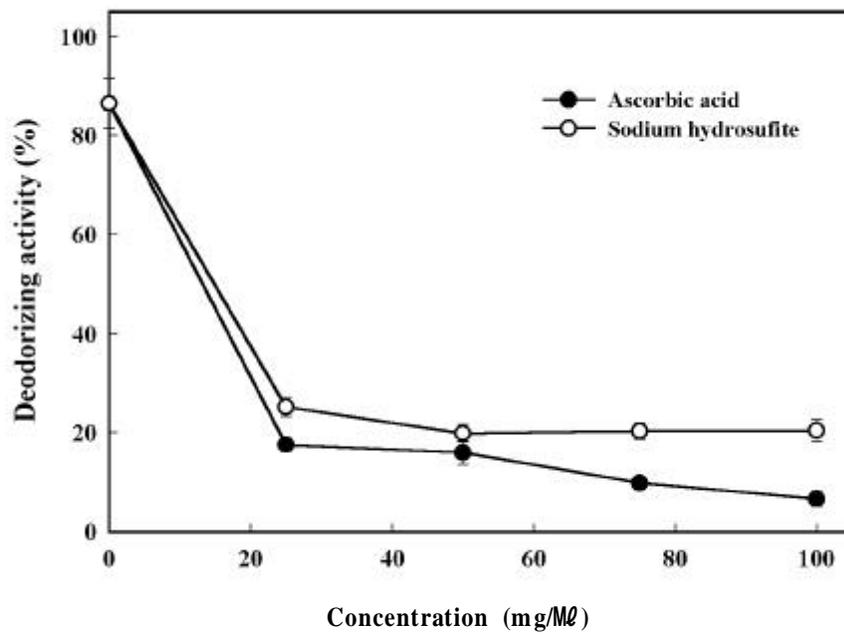


Fig. 8. Effect of reducing agents on the deodorizing activity of polyphenol oxidase from apples against methyl mercaptan.

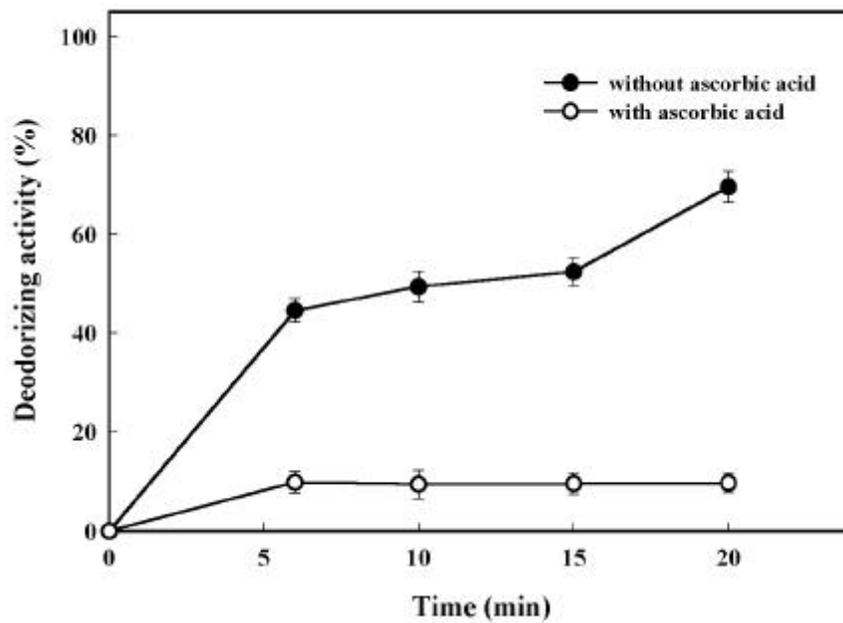


Fig. 9. Effect of ascorbic acid on the deodorizing activity of polyphenol oxidase from apples against methyl mercaptan.

Ascorbic acid 가

methyl mercaptan 가 ascorbic acid

가 methyl mercaptan

ascorbic acid 가 polyphenol

oxidase methyl mercaptan ,

ascorbic acid가 methyl mercaptan

o-quinone

5. 3.

polyphenol oxidase methyl mercaptan

Fig. 10 .

(O₂)가

가 helium gas

가 . Mcevely et al.(1992) *o*-diphenols polyphenol

oxidase, O₂ *o*-quinones

ascorbic acid sulfites

helium gas

o-diphenols 가

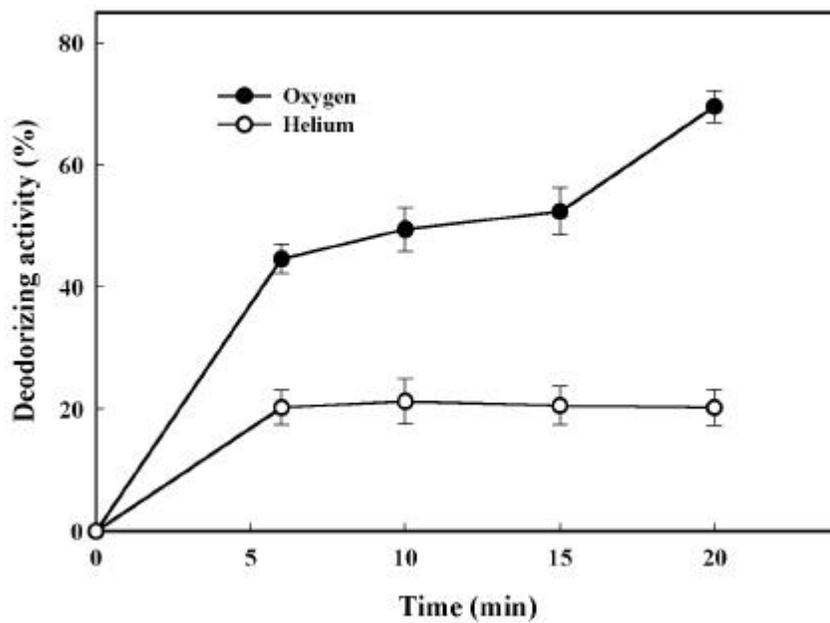


Fig. 10. Effect of oxygen on the deodorizing activity of polyphenol oxidase from apples against methyl mercaptan.

5. 4.

polyphenol oxidase methyl mercaptan

10^{-1} M

Table 3

Polyphenol oxidase 가

L-tyrosine 가 ,

L-DOPA gallic acid 가

chlorogenic acid

catechol Cu^{2+} , Fe^{3+} 가 ,

Ca^{2+} , Mg^{2+} . Kim et al.(1983)

polyphenol oxidase 10^{-1} mM 가 ,

Cu^{2+} , Mn^{2+} , Fe^{3+} 가 Na^+ , Hg^{2+} ,

Co^{2+} .

Kim et al.(1981) polyphenol oxidase Cu^{2+}

polyphenol oxidase (10^{-2}

1.0 mM) Cu^{2+} 가 , (10mM)

Cu^{2+}

가 Cu^{2+} 가 polyphenol oxidase cofactor

polyphenol oxidase

가 chlorogenic

acid Cu^{2+} Fe^{2+} 가 93.5, 88.7%

Table 3. Effect of metals and polyphenols on the deodorizing activity of polyphenol oxidase from apples against methyl mercaptan

Substrate	Metal Salt (10 ⁻¹ M)	Activity (unit)	RA ^a (%)	DA ^b (%)
L-tyrosine	None	0	0	0
	CuCl ₂	0	0	0
	FeCl ₃	0	0	0
	CaCl ₂	0	0	0
	MgCl ₂	0	0	0
Chlorogenic acid	None	76.0	190.0	82.5
	CuCl ₂	107.9	269.8	93.5
	FeCl ₃	91.2	225.5	88.7
	CaCl ₂	76.0	100.0	81.5
	MgCl ₂	85.1	212.8	83.7
Catechol	None	40.0	100.0	34.7
	CuCl ₂	55.2	138.0	48.5
	FeCl ₃	48.4	121.0	40.4
	CaCl ₂	39.2	98.0	34.5
	MgCl ₂	42.0	105.0	36.5
L-DOPA ^c	None	6.4	16.0	12.6
	CuCl ₂	7.0	17.5	14.7
	FeCl ₃	6.9	17.3	13.5
	CaCl ₂	6.3	15.8	11.9
	MgCl ₂	6.6	16.5	12.8
Gallic acid	None	1.5	3.8	11.9
	CuCl ₂	1.7	4.3	12.8
	FeCl ₃	1.6	4.0	12.6
	CaCl ₂	1.5	3.8	11.7
	MgCl ₂	1.5	3.8	11.3

^aRA means relative activity,

^bDA means deodorizing activity,

^cL-DOPA means L-3,4-dihydroxyphenylalanine.

가
Cu²⁺ Fe³⁺ polyphenol
oxidase cofactor o-quinone

5. 5. Ethylenediaminetetraacetic acid(EDTA)

polyphenol oxidase methyl mercaptan
EDTA
25mM EDTA 가 Table 4
EDTA 가 , polyphenol oxidase
EDTA 가
(1984) polyphenol
oxidase 100 mM EDTA 가 가 62%
chelator EDTA가 polyphenol oxidase
prosthetic group Cu o-quinone

Table 4. Effect of ethylenediaminetetraacetic acid(EDTA) on the deodorizing activity of polyphenol oxidase from apples against methyl mercaptan

	Substrate (10mM)	Activity (unit)	RA ^a (%)	DA ^b (%)
Without EDTA	L-tyrosine	0	0	0
	Chlorogenic acid	75.0	208.3	82.5
	Catechol	36.0	100.0	36.1
	L-DOPA ^c	6.5	18.1	13.5
	Gallic acid	1.4	3.9	13.1
With EDTA	L-tyrosine	0	0	0
	Chlorogenic acid	65.5	181.9	74.2
	Catechol	30.0	83.3	28.5
	L-DOPA	4.7	13.1	10.5
	Gallic acid	0	0	0

^aRA means relative activity,

^bDA means deodorizing activity,

^cL-DOPA means L-3,4-dihydroxyphenylalanine.

polyphenol oxidase

, pH, , , EDTA .

.

1. methyl mercaptan

, 가 가

가

10mg/ml 72.5% .

2. Catechol polyphenol oxidase pH

, polyphenol oxidase pH 6.5,

30°C .

3. polyphenol oxidase

가 ,

가 methyl mercaptan

가 .

4. polyphenol oxidase
 polyphenol oxidase
 , chlorogenic acid catechol
 81.3% 36.8% .

5. polyphenol oxidase
 pH pH 5.5 20% ,
 80% .

6. polyphenol oxidase ascorbic
 acid sodium hydroxide 가
 30% .

7. 가 , 20% .

8. polyphenol oxidase Cu^{2+} Fe^{3+} 가
 가 ,
 Ca^{2+} Mg^{2+} 가 가 .

9. polyphenol oxidase

EDTA 가

.

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